



understanding new energies

## H<sub>2</sub> Lab - a Combined Training Platform

Michael Dietrich  
Dr. Dmitry Kushnikovskiy  
leXsolar GmbH



## About us

- Established in 2003 as a spin-off from the Technical university of Dresden/ Germany
- Worlddidac Award Winner
- One of the leaders in education in renewable energies
- Active in 60+ countries with over 50 partners



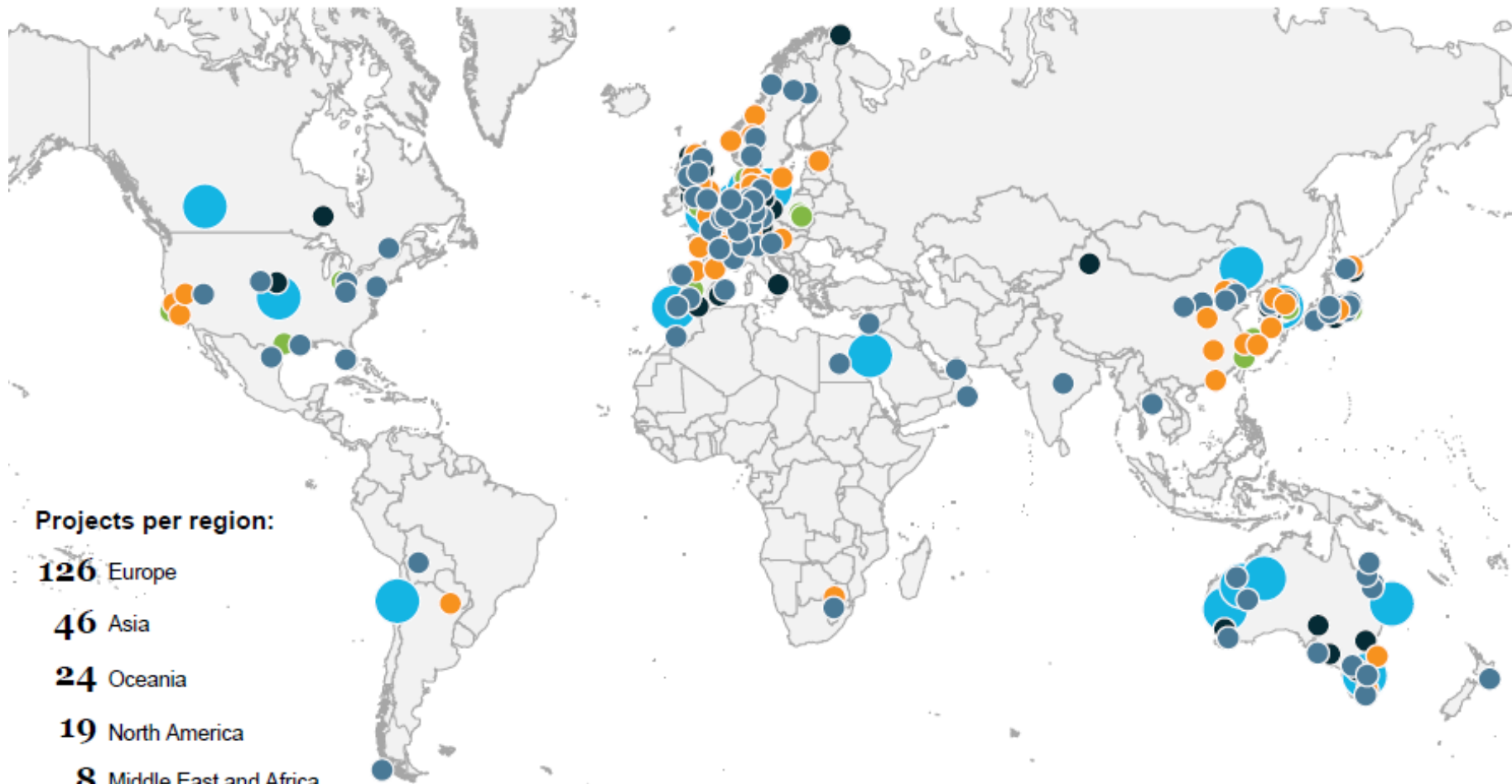
# leXsolar areas of expertise



# References



# Global Hydrogen Projects



**Projects per region:**

- 126** Europe
- 46** Asia
- 24** Oceania
- 19** North America
- 8** Middle East and Africa
- 5** Latin America

**228** announced projects

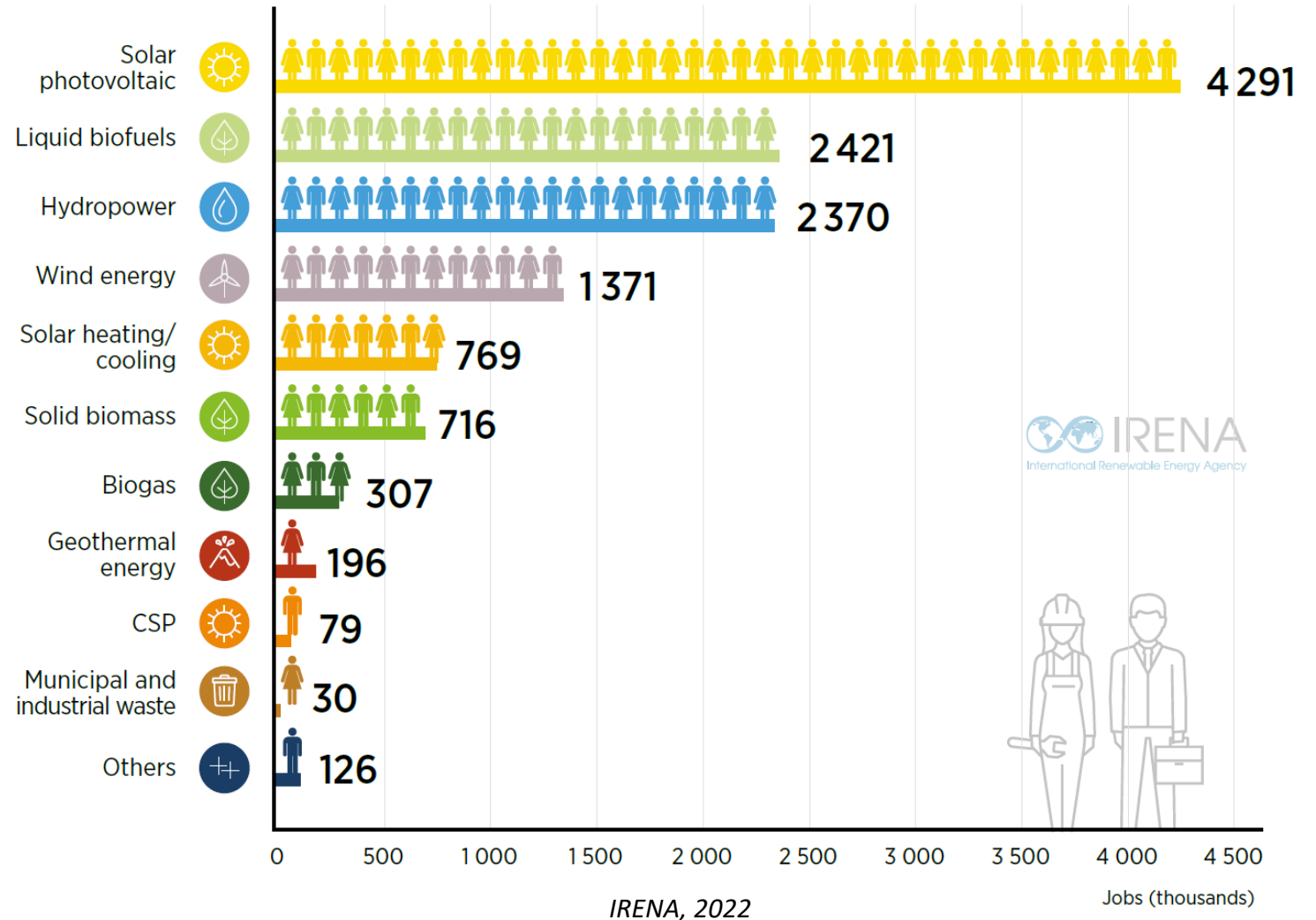
*Hydrogen Council, McKinsey & Company, Hydrogen Insights Report 2021*

- 17**  
Giga-scale production: renewable H<sub>2</sub> projects >1GW and low-carbon H<sub>2</sub> projects >200 kt p.a.
- 90**  
Large-scale industrial usage: refinery, ammonia, power, methanol, steel, and industry feedstock
- 53**  
Transport: trains, ships, trucks, cars and other hydrogen mobility applications
- 45**  
Integrated H<sub>2</sub> economy: cross-industry, and projects with different types of end-uses
- 23**  
Infrastructure projects: H<sub>2</sub> distribution, transportation, conversion, and storage



# Jobs in Renewable Energy Sector

- Doubled since 2011
- Ca. 38 million are expected globally by 2030
- 134 million in the energy sector
- Asia accounted for 63% of total jobs in renewables globally



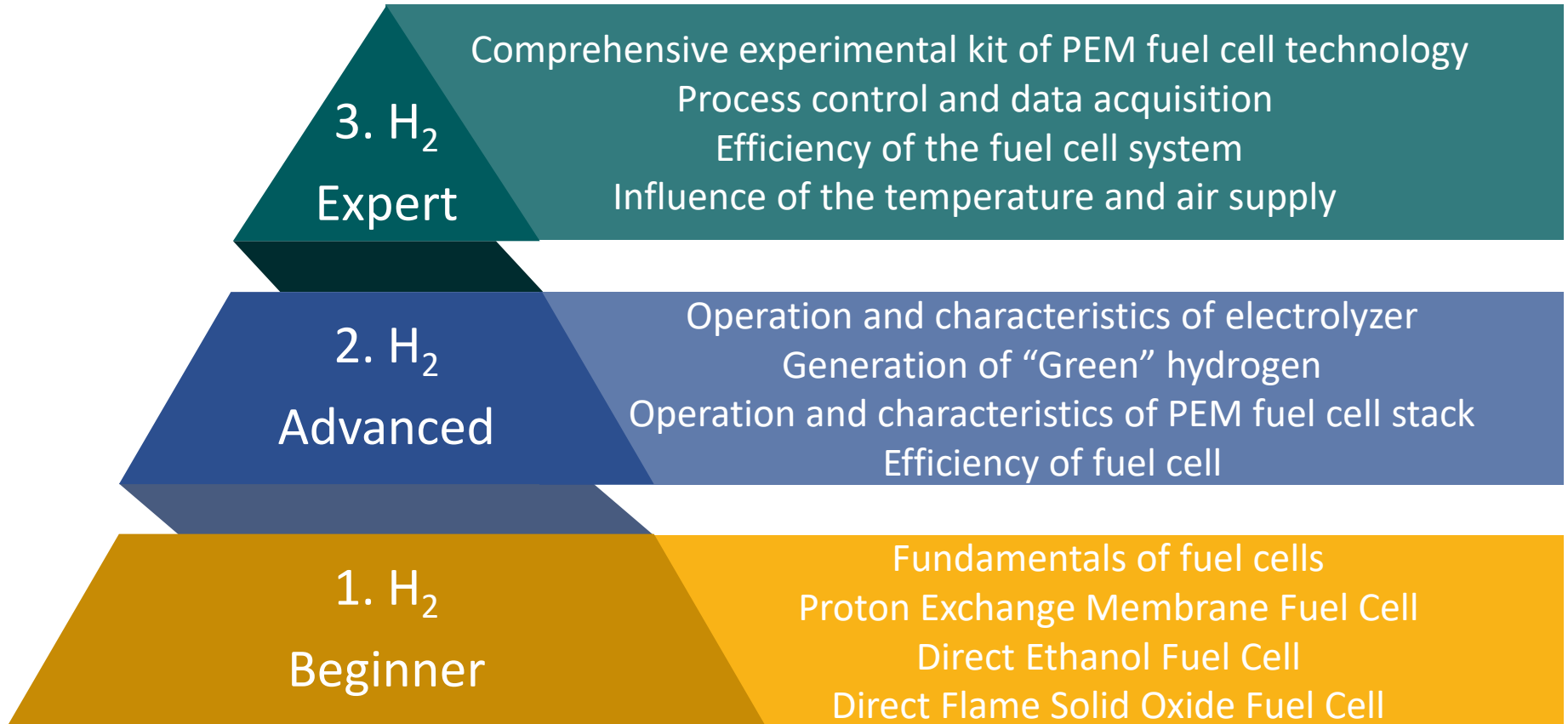


## Module 0

- History of Fuel Cell Technologies
- H<sub>2</sub> Production
- H<sub>2</sub> Transport
- H<sub>2</sub> Storage
- H<sub>2</sub> Applications



## leXsolar Experimental Kits





- Certification
- Blockchain credentials
- Issuer verification
- Certificate validation
- Skill pass

**H<sub>2</sub> SPECIALIST - NUMBER 2657**

**CERTIFICATE**

THIS TRAINING CERTIFICATE IS PRESENTED TO:

**ERIK MÜLLER**

Who has been successfully evaluated  
on the experimental and theoretical course of leXsolar Specialist:

**H<sub>2</sub> Specialist**

Powered by:

Michael Dietrich  
leXsolar CEO  
01 / 08 / 2022  
Date completed

leXsolar  
leXsolar GmbH

**HYTEC POWER INC.**  
Certification center

Partner of: worlddidac ASSOCIATION | didacta (DIDACTA) | GREEN Solar Academy

**H<sub>2</sub> Certificate Specialist**

**H<sub>2</sub> Microcertificate Expert**  
with 1223 leXsolar H<sub>2</sub> Expert for Technicians

**H<sub>2</sub> Microcertificate Advanced**  
with 1222 leXsolar H<sub>2</sub> Professional for TVET Users

**H<sub>2</sub> Microcertificate Elemental**  
with 1224 eXsolar H<sub>2</sub> Study-Log for Engineers

**H<sub>2</sub> LAB Online learning course**

1. History of Fuel Cell technologies
2. H<sub>2</sub> Production
3. H<sub>2</sub> Transport
4. H<sub>2</sub> Storage
5. H<sub>2</sub> Applications





<b>Target group</b>	TVET (Technical and Vocational Education and Training)
<b>ISCED Level</b>	5 - Short-cycle tertiary education
<b>Age group</b>	16 – 50+
<b>Number of students</b>	20-30 students per classroom
<b>Students per equipment</b>	6-12 Students can work simultaneously with experimental kits
<b>Focus of lab-equipment</b>	Green Energy Education for TVET
<b>Subjects:</b>	Renewable Energies, Environmental and Electrical Engineering, STEM, Chemical
<b>Quality standard:</b>	leXsolar is an official member of the Worlddidac Association and Didacta e.V. Germany

**7 AFFORDABLE AND CLEAN ENERGY**



**4 QUALITY EDUCATION**



**13 CLIMATE ACTION**



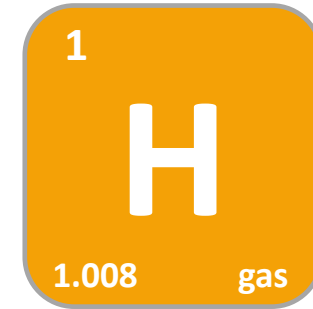
**8 DECENT WORK AND ECONOMIC GROWTH**





# Module 0: Fundamentals of hydrogen technologies

- **Previous knowledge:**
  - School level STEM
- **Learning objectives:**
  - Hydrogen safety
  - Basics of hydrogen economy
  - Current and emerging methods for producing hydrogen
  - Applications of hydrogen and fuel cells
  - Options for storing and transporting hydrogen
- **Duration:**
  - 3 units
  - 4 – 8 hours (self-paced),
  - up to 16 hours (as a part of the guided lessons)





# Module 0: Fundamentals of hydrogen technologies

## Production of Hydrogen

Colors of hydrogen

Hydrogen emits just water when burned - but creating it can be very carbon intensive.

Various ways to lessen this impact have been developed one is the assigning of colors to the different types:

Continue

H2- Basics and production of hydrogen

## Transport and Storage

Transporting hydrogen via road.

Today, hydrogen normally is transported from the point of production to the point of use via pipeline and over the road in cryogenic liquid tanker trucks or gaseous tube trailers.

Due to the achievable transport volume road transport is used primarily for smaller quantities and local distribution.

Continue

H2- Basics and production of hydrogen

## Application

There are several types of fuel cells, that vary based on their operating temperature, materials, and design. PEMFC is currently the most researched and tested type in industry and academia, while AFC, PAFC, and MCFC have declined in importance due to technical issues during field tests and a lack of confidence in cost reduction.

Continue

H2- Basics and production of hydrogen

- Hydrocarbon Reforming
  - Steam Reforming
  - Partial Oxidation
- Hydrocarbon Pyrolysis
- Biological Processes
- Electrolysis
  - Alkaline
  - Solid Oxide Electrolyser Cell
  - Proton Exchange Membrane
- Other methods

- Chemical-based storage
  - Adsorbent
  - Liquid organic
  - Hydride
- Physical-based storage:
  - Liquid storage
  - Gas storage
- Hydrogen transport

- Hydrogen as energy carrier
- Hydrogen as fuel
  - Portable
  - Stationary
  - Mobility
- Hydrogen in fuel cells
  - Proton Exchange Membrane
  - Alkaline
  - Phosphoric Acid
  - Molten Carbonate
  - Solid Oxide
  - Direct Alcohol



# Module 1: H<sub>2</sub> Beginner

## Previous knowledge:

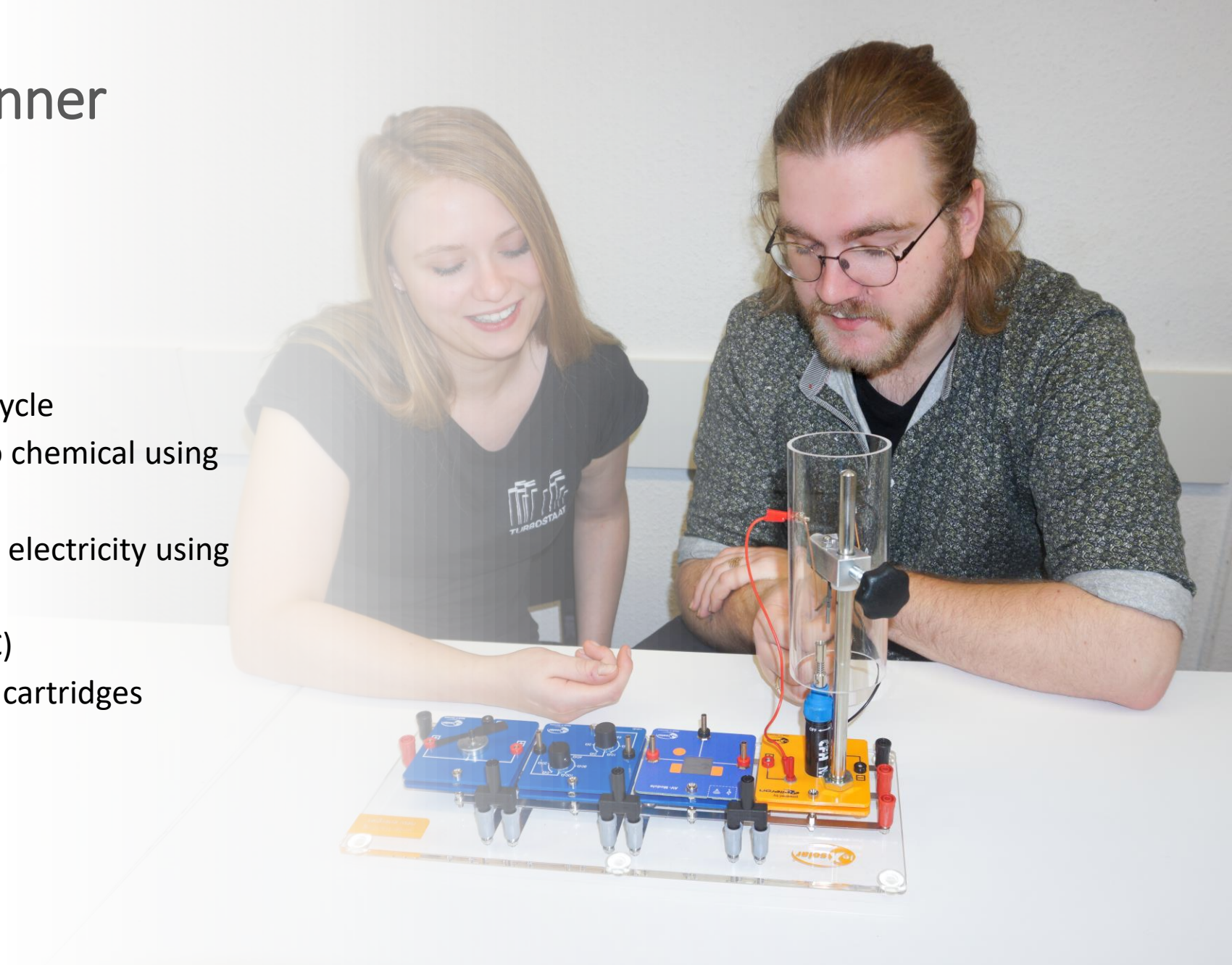
- Not required

## Learning objectives:

- Basics of a solar-hydrogen energy cycle
- Conversion of electrical energy into chemical using electrolyser
- Conversion of chemical energy into electricity using fuel cells
- Types of fuel cell (PEM, EtOH, SOFC)
- Hydrogen storage in metal-hydride cartridges

## Duration:

- 4 main units; 8 - 12 hours

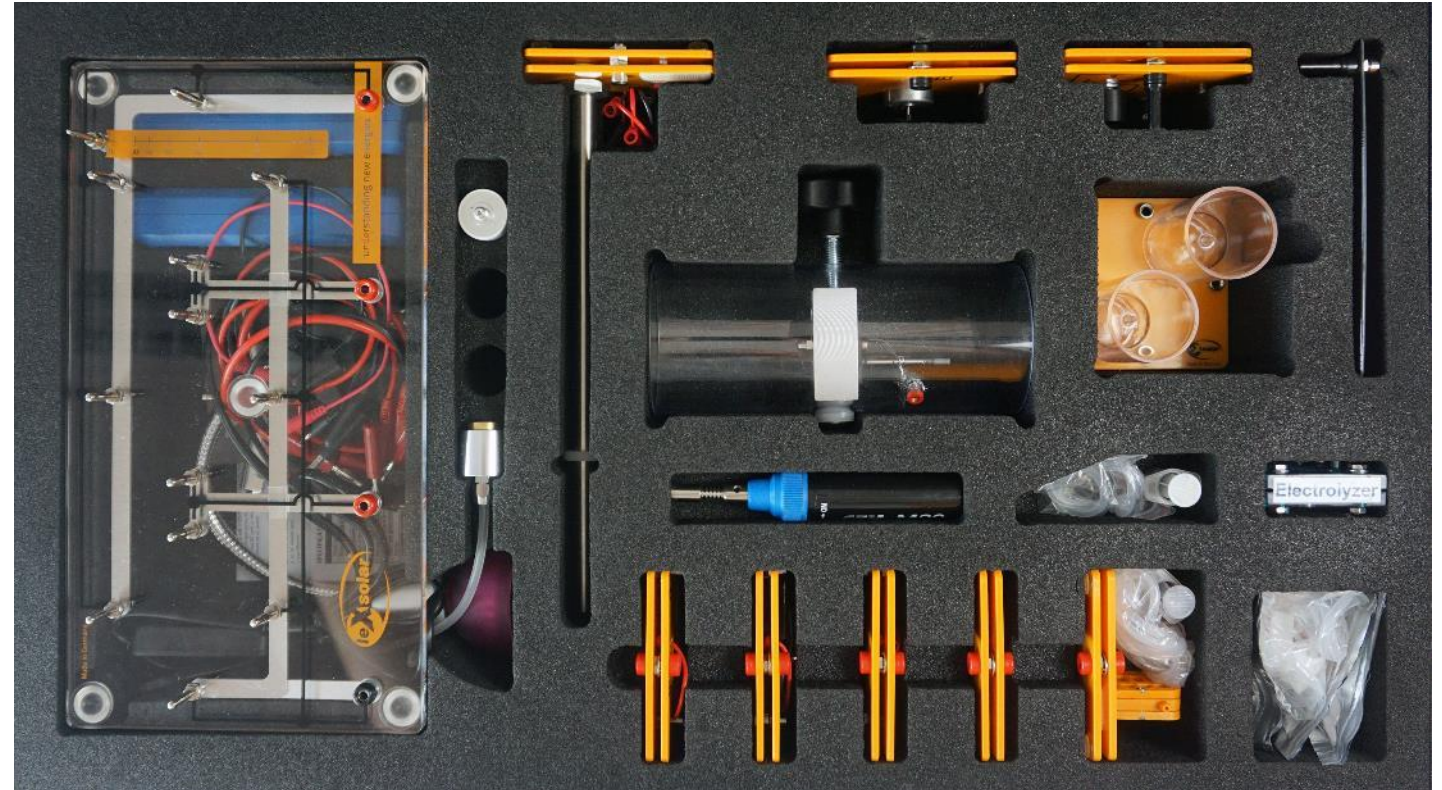






# H<sub>2</sub> Beginner: Components

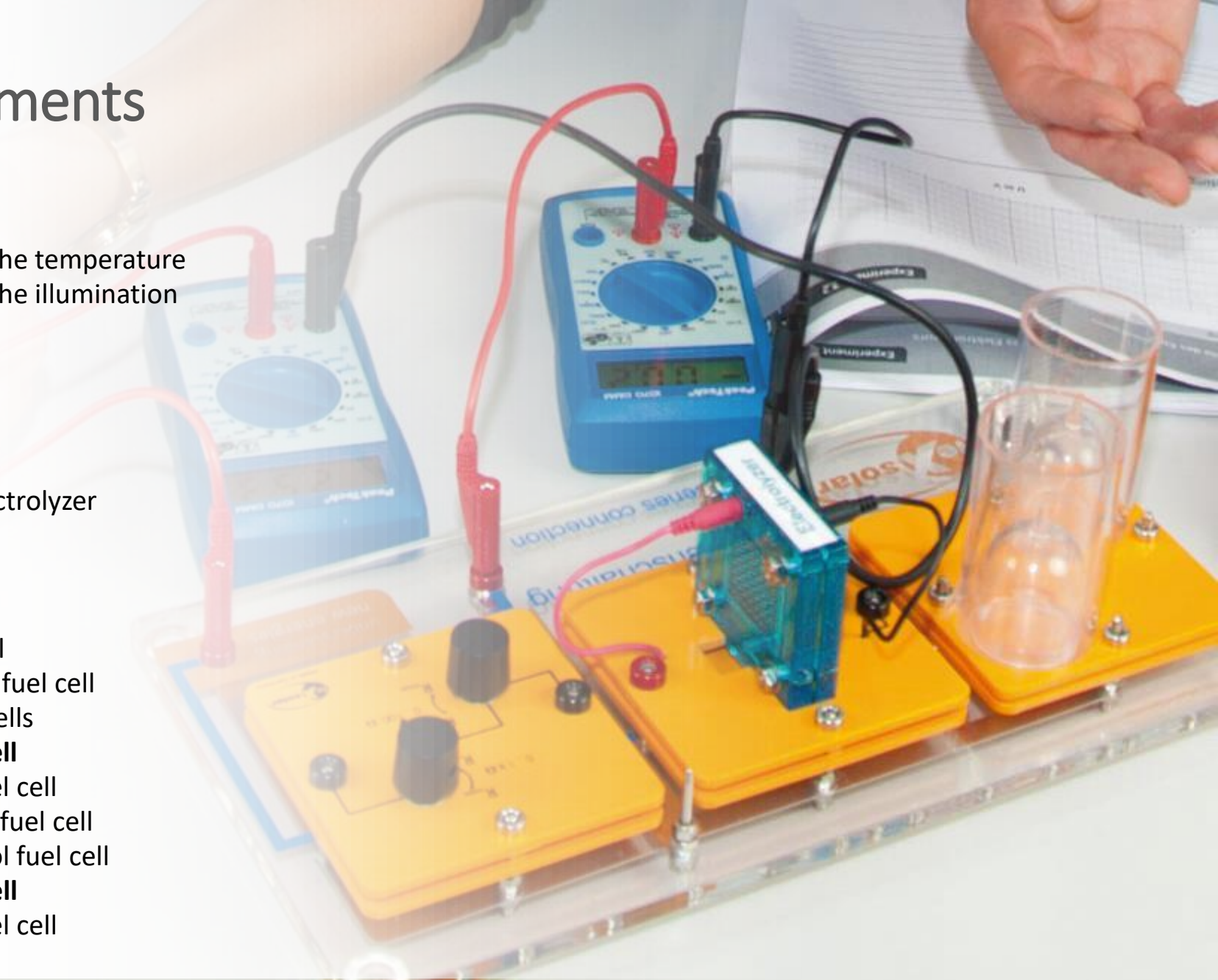
- Base unit
- Potentiometer module
- Motor module
- Solar module 2.5 V, 420 mA
- H<sub>2</sub> Storage
- Gas storage module
- 3x PEM-Fuel Cell Module
- Electrolyzer module 2.0
- Ethanol fuel cell
- SOFC fuel cell
- Fuel cell stand
- Gas burner
- Lamp with table clamp
- 2x Digital multimeter
- 2x Test lead black 25 cm
- 2x Test lead red 25 cm
- Test lead black 50 cm
- Test lead red 50 cm
- Valve for H<sub>2</sub> Storage
- Silicone hose 4 mm (o.d.)





# H<sub>2</sub> Beginner: Experiments

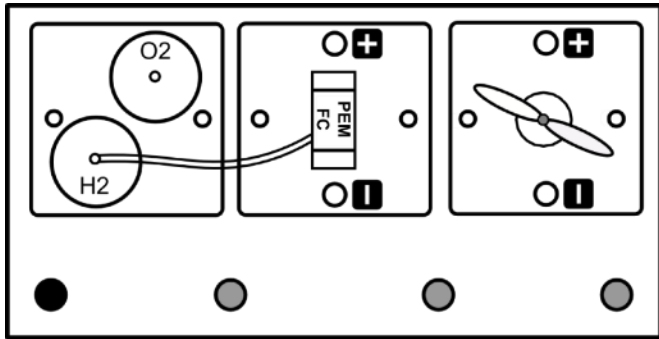
- **Basic experiments:**
  - I-V characteristics of the solar cell
  - Dependence of the solar cell power on the temperature
  - Dependence of the solar cell power on the illumination intensity
- **Experiments with electrolyzer:**
  - Generation of “green” hydrogen
  - I-V characteristics of the electrolyzer
  - Faraday and energy efficiency of the electrolyzer
- **Experiments with fuel cell:**
  - **Properties of a PEM fuel cell**
  - I-V characteristic curve of a PEM fuel cell
  - Faraday- and energy efficiency of a PEM fuel cell
  - Series and parallel circuits of PEM fuel cells
  - **Working principles of an ethanol fuel cell**
  - I-V characteristic curve of an ethanol fuel cell
  - Temperature dependence of an ethanol fuel cell
  - Concentration dependence of an ethanol fuel cell
  - **Working principles of solid oxide fuel cell**
  - I-V characteristic curve of an ethanol fuel cell



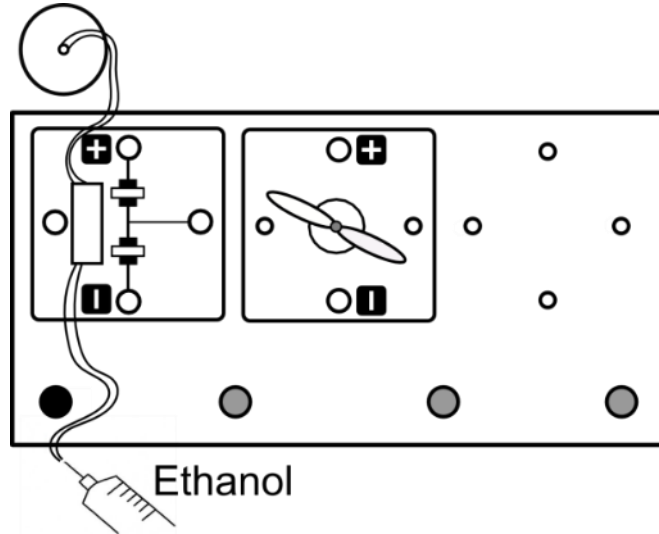


# H<sub>2</sub> Beginner: Working principle of the fuel cells

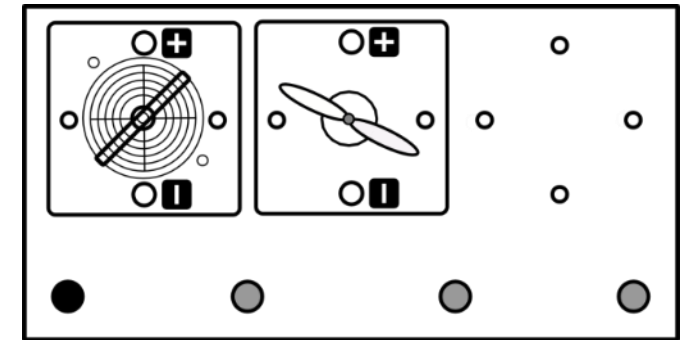
Investigate and compare the working principle of PEM, EtOH, and SOFC fuel cells



- Base unit
- Electrolyser or H<sub>2</sub> storage
- PEM fuel cell
- Motor module



- Base unit
- Ethanol fuel cell
- Ethanol
- Syringe
- Motor module



- Base unit
- SOFC fuel cell
- Gas burner
- Motor module





## Module 2: H<sub>2</sub> Advanced

### Previous knowledge:

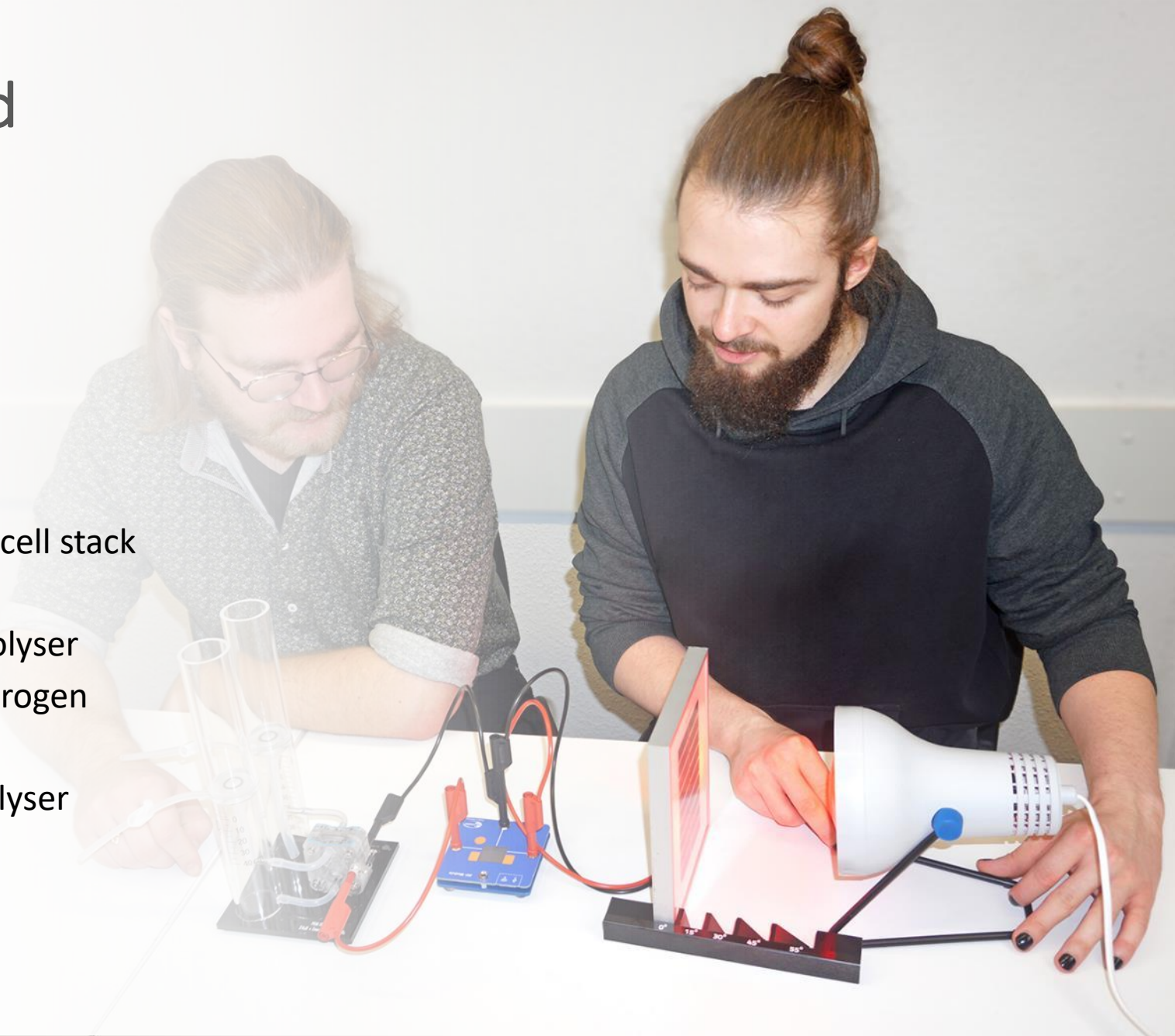
- Fundamentals of electrolysis
- Working principle of PEM fuel cells

### Learning objectives:

- Basic properties of the solar cell
- Operation and properties of the PEM fuel cell stack (1-5 cells)
- Operation and characteristics of an electrolyser
- Solar-hydrogen energy cycle for green hydrogen generation
- Efficiency of the fuel cell stack and electrolyser

### Duration:

- 3 main units; 8 - 12 hours







## Module 2: H<sub>2</sub> Advanced

- Double cell electrolyser
- Fuel cell stack (1-5 cells)
- Model car
- Base unit
- Solar cell module 5,2 V with stand
- Infrared lamp
- Potentiometer
- AV module
- Power module
- Blower fan
- H<sub>2</sub> storage (metal hydride, 30 bar, 10 l)
- One-step pressure regulator
- Adapter 2 mm/4 mm
- Short circuit plugs
- Safety cables (4 mm), adapters
- Aluminum case





# H<sub>2</sub> Advanced: Experiments

- **Basic experiments:**

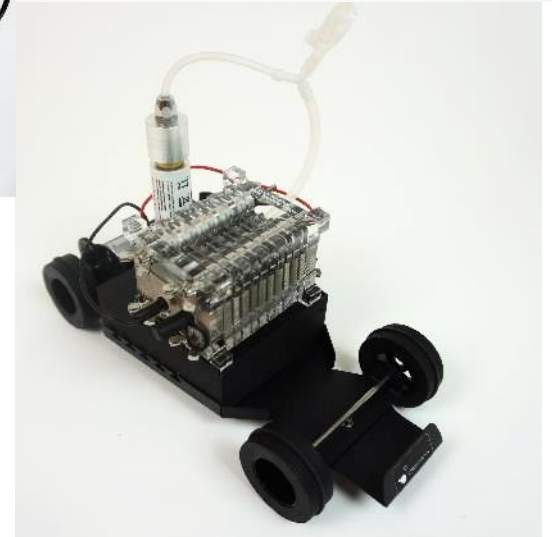
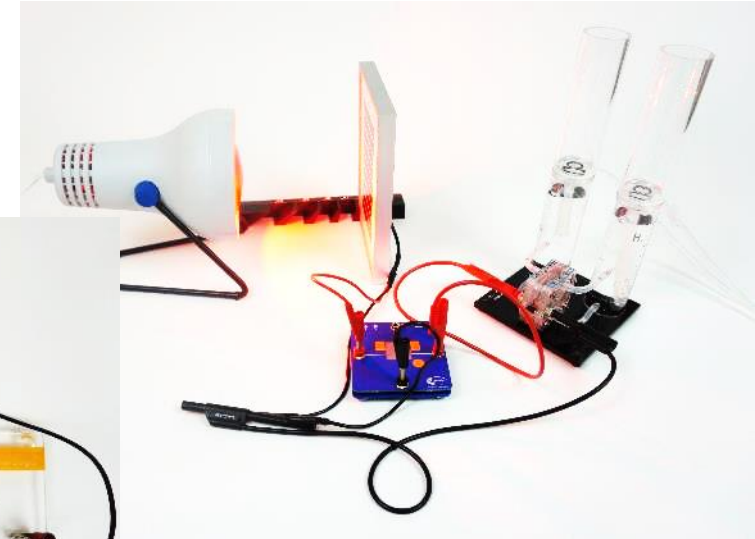
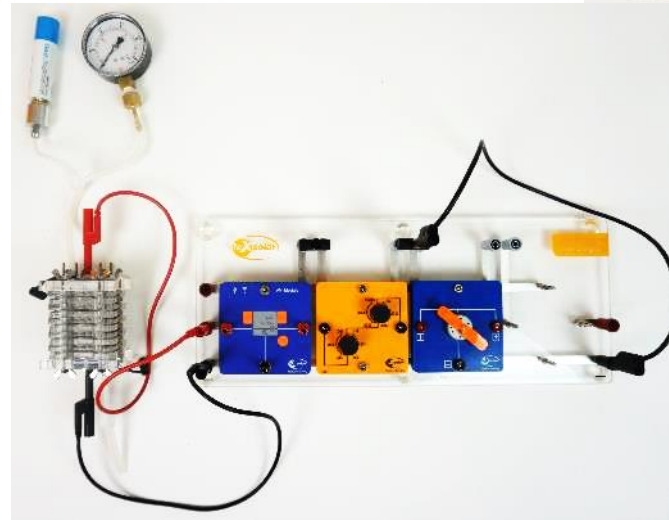
- I-V characteristics of the solar cell
- Dependence of the solar cell power on the temperature
- Dependence of the solar cell power on the illumination intensity

- **Experiments with electrolyzer:**

- Properties of the electrolyzer:
- I-V characteristics of the electrolyzer
- **Solar-powered generation of hydrogen**
- Faraday and energy efficiency of the electrolyzer

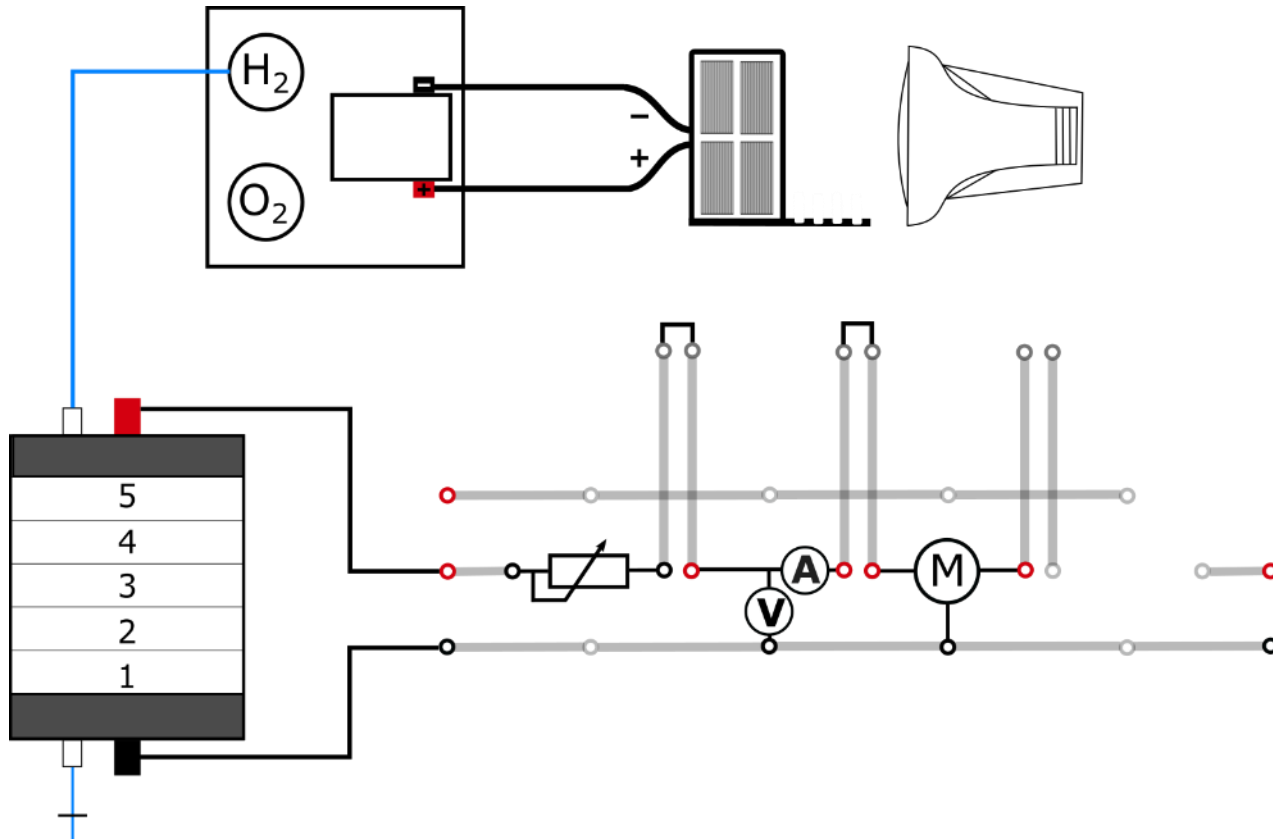
- **Experiments with fuel cell:**

- **Operation of the consumer with the 5-cell stack**
- Examination of the single cell compared to the fuel cell stack
- Operation of the fuel cell stack with and without fan
- Faraday and energy efficiency of the fuel cell stack
- Operation of the model car with a fuel cell stack
- Hydrogen consumption of the fuel cell



# H<sub>2</sub> Advanced: Green Hydrogen Production

Direct utilization of produced “green” hydrogen in PEM fuel cell stack



## Equipment:

- Base unit
- Solar module + base for solar panel
- Lamp
- AV-Module
- Potentiometer module Electrolyser
- Fuel cell stack
- Cables and adapters

## Optional:

- Power module, 4V (for “brown” hydrogen generation)



# Module 3: H<sub>2</sub> Expert

## Previous knowledge:

- Working principle of the PEM fuel cell
- Efficiency of the PEM fuel cell

## Learning objectives:

- PEM fuel cell stack
- Process control and efficiency of the fuel cell system
- Operating modes of the fuel cell system
- Recognizing and eliminating errors
- Hydrogen consumption

## Duration:

- 1 main unit; 6-10 hours







# H<sub>2</sub> Expert: Components

## Fuel cell stack:

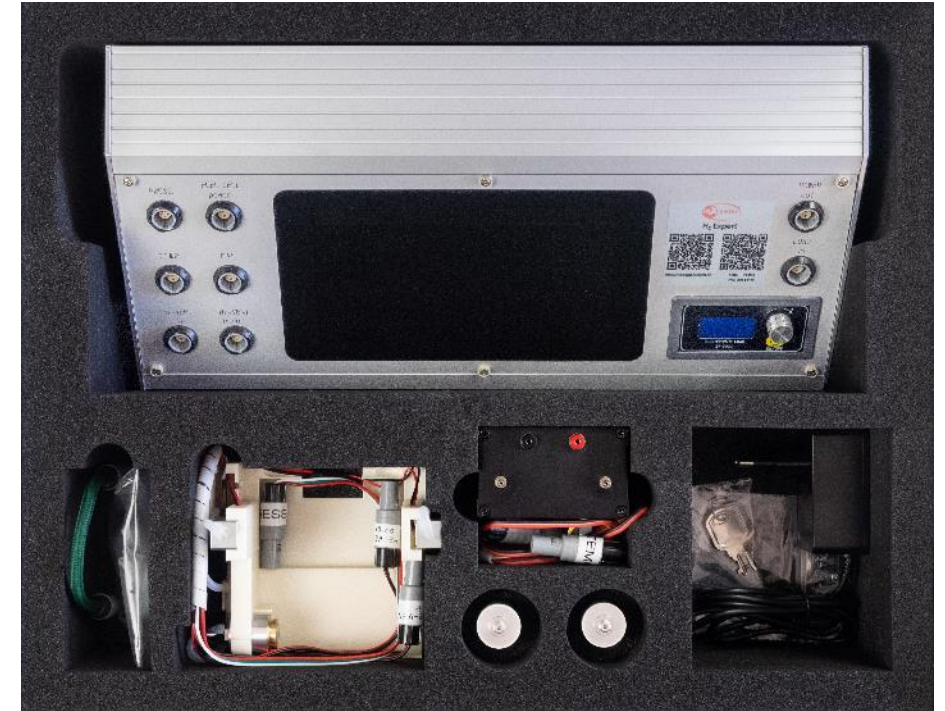
- Type: Proton Exchange Membrane (PEM)
- Nominal power: 20 W (7.6 V @ 2.6 A)
- Open circuit voltage: 12 V
- Start-up time: ≤ 30 s (25 °C)
- Maximal stack temperature: 55 °C
- H<sub>2</sub> pressure: 0.4 - 0.6 bar

## Controller:

- Control unit for the fuel cell
- Data acquisition (cell temperature, H<sub>2</sub> pressure, voltage, current, power)
- Integrated webserver with control dashboard and data logger
- Error management
- Export of measured data as .csv

## Electronic load:

- Input voltage: 1 – 30 V
- Discharge current: 0 - 5 A, adjustable in 0.01 A
- Power supply: 12 V
- Maximum power: 35 W





## H<sub>2</sub> Lab: Safety

- Small volume of produced hydrogen and oxygen (max. 80 ml)
- Relatively low-pressure metal hydride hydrogen storage (30 Bar, max. 10 L)
- Hydrogen generator for recharging metal-hydride cartridges
- Direct alcohol fuel cell with non-toxic ethanol as a fuel
- Solid oxide fuel cell does not require an external heater
- Can be used in almost every laboratory and classroom with adequate ventilation



## Contact

leXsolar GmbH  
Strehleener Straße 12-14  
01069 Dresden  
Germany

Tel: + 49 351 - 47 96 56 101  
Fax: + 49 351 - 47 96 56 111  
e-mail: [info@leXsolar.com](mailto:info@leXsolar.com)



[www.leXsolar.com](http://www.leXsolar.com)